

Ammonia handling at DTU Energy

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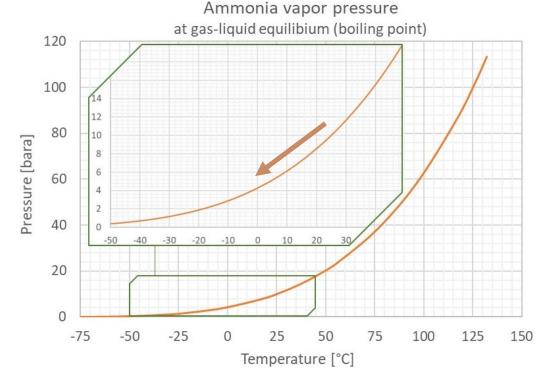


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Challenges: NH₃ bottles cooling down



The NH₃ pressure in the bottle drops \longrightarrow The ammonia supply stops if p < 3 bar (min pressure for the MFC to work)

SOLUTION → heating blankets to keep the NH₃ bottle warm enough



Challenges: condensation

The heating blankets could get too warm



The NH₃ T and p in the bottles become too high



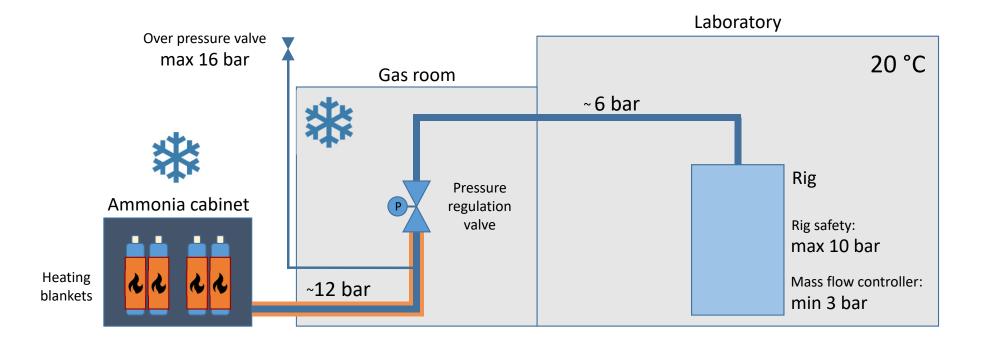
- $1_{\text{the rig}}$ safety alarm triggers if p > 10 bar (max pressure allowed for the ammonia in the rig)
- 2_NH₃ can condensate in the pipes of the lab if they are at a T lower than the one of the bottles Risky for the MFC
- 3_ the NH_3 flowing through the MFC expands a lot (~9 bar \rightarrow ~1 bar), cooling down significantly (when it happened to us, the NH_3 pipe was covered with frost) and condensing

 Too cold gas (or even liquid) to the stack

SOLUTION \rightarrow pressure regulation valve, to keep the pressure (and temperature) inside the pipes at acceptable values

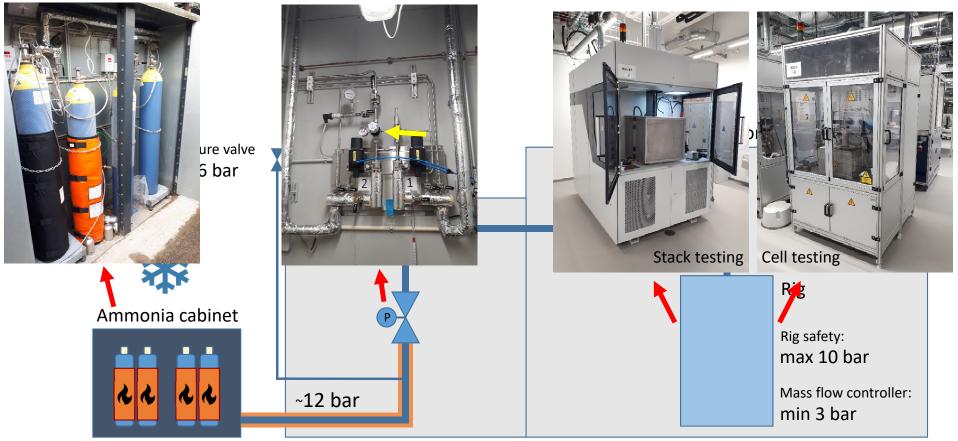


NH₃ supply line



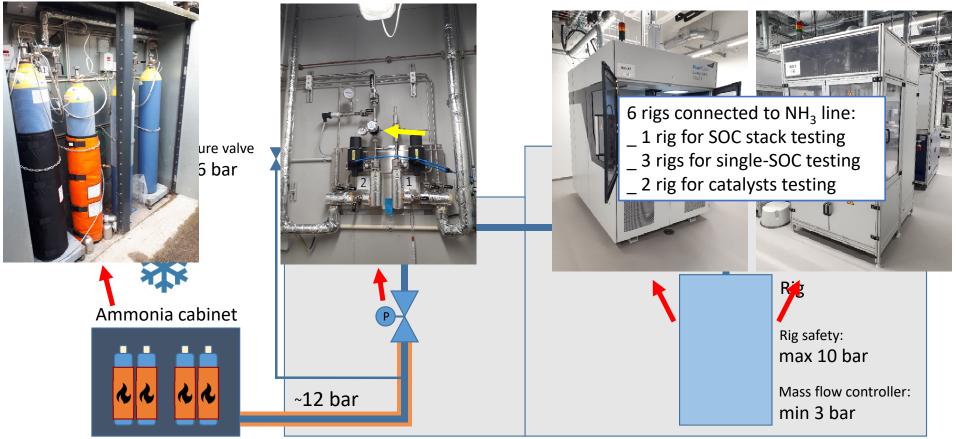


NH₃ supply line





NH₃ supply line





Failing parts

Broken magnetic valve



Mass flow controller



Ammonia is quite aggressive against some materials, so it is important to install valves and mass flow controllers made of materials suitable for such gas



Safety measures

protections to use while working with the ammonia gas line



Mask and air bottle



Protective apron

- _ ammonia sensors placed on the lab ceiling and inside the rigs to detect leakages
- _ over pressure valve, to discharge ammonia to the atmosphere if the pressure is dangerously high
- _ door safety alarms in the rig: if NH₃ is used and the door of the rig opens, the flow of NH₃ is stopped automatically



Summary

- ➤ powerful and well-functioning ammonia infrastructure that allows a stable supply for testing of cells, stacks and catalysts, even with high flow rates
- reliable instruments (valves and MFC) that allow long-term testing
- > strict safety measures for a secure work environment